

The 81st Fujihara seminar
2024.6.3. 16:15-16:35



Quantitative Analysis Around Crystallization Phenomena via Molecular Electron Microscopy

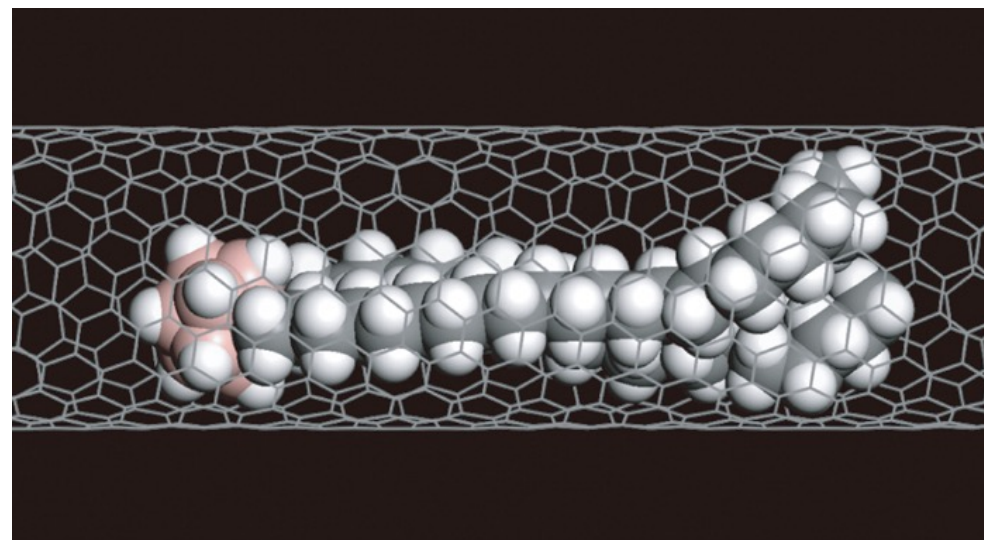
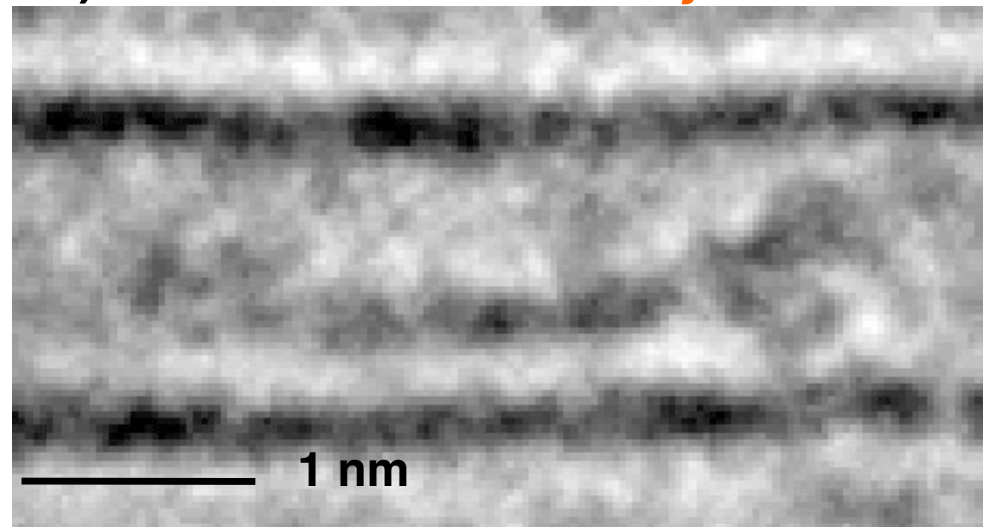
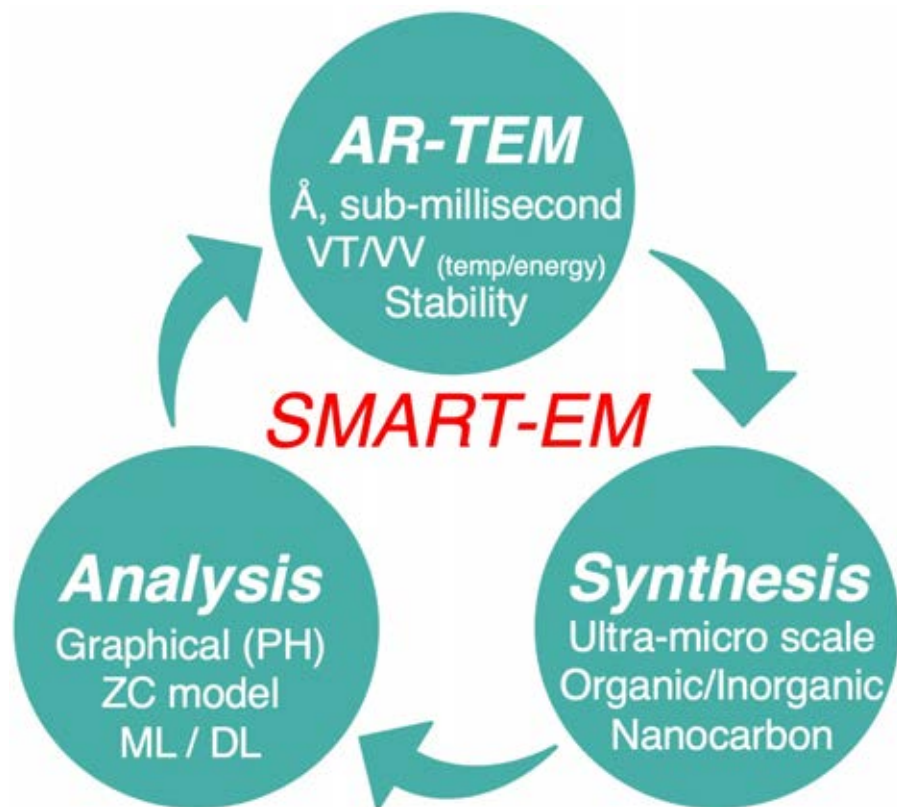
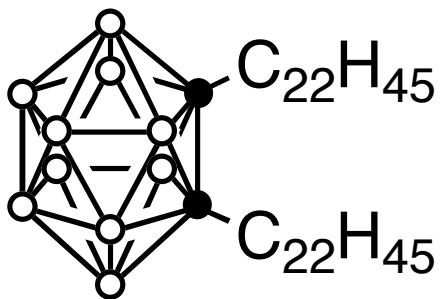
Takayuki NAKAMURO

“Molecular Technology Innovation”
Presidential Endowed Chair



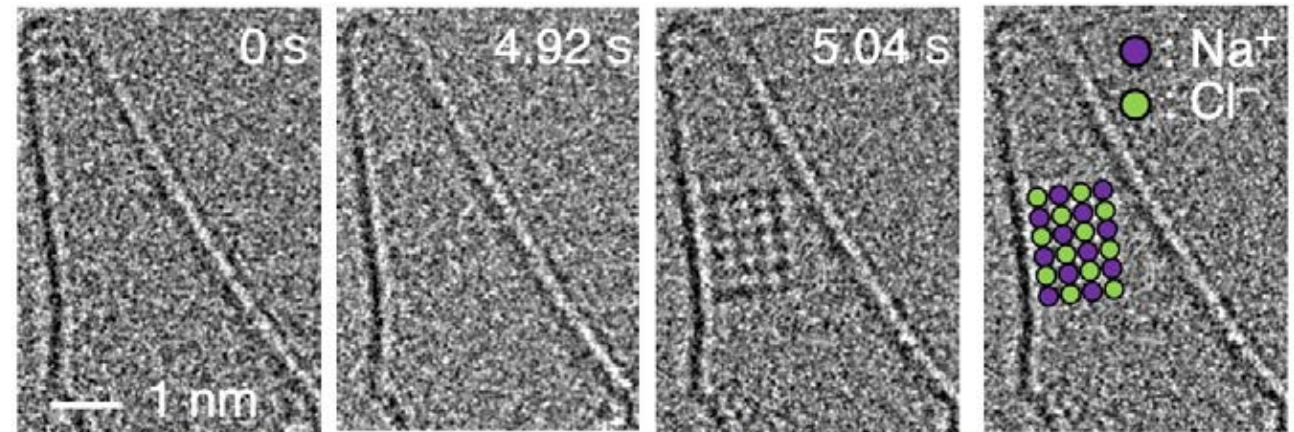
The First Movie of a Single Molecule in Motion

Single-Molecule, Atomic-Resolution Time-resolved
Electron Microscopy (SMART-EM) or *Cinematic Chemistry*

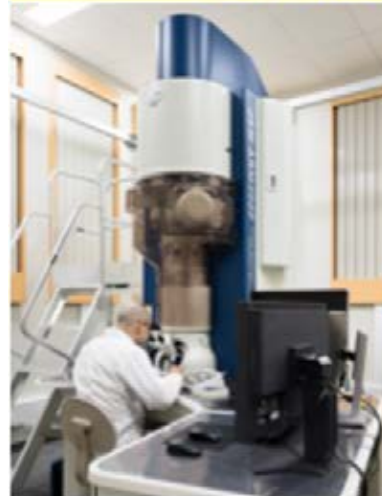


K. Suenaga, H. Isobe, E. Nakamura *et al. Science* **2007**, 316, 853.

Capturing the Moment of Crystallization



Chaotic • Nucleation • Growth



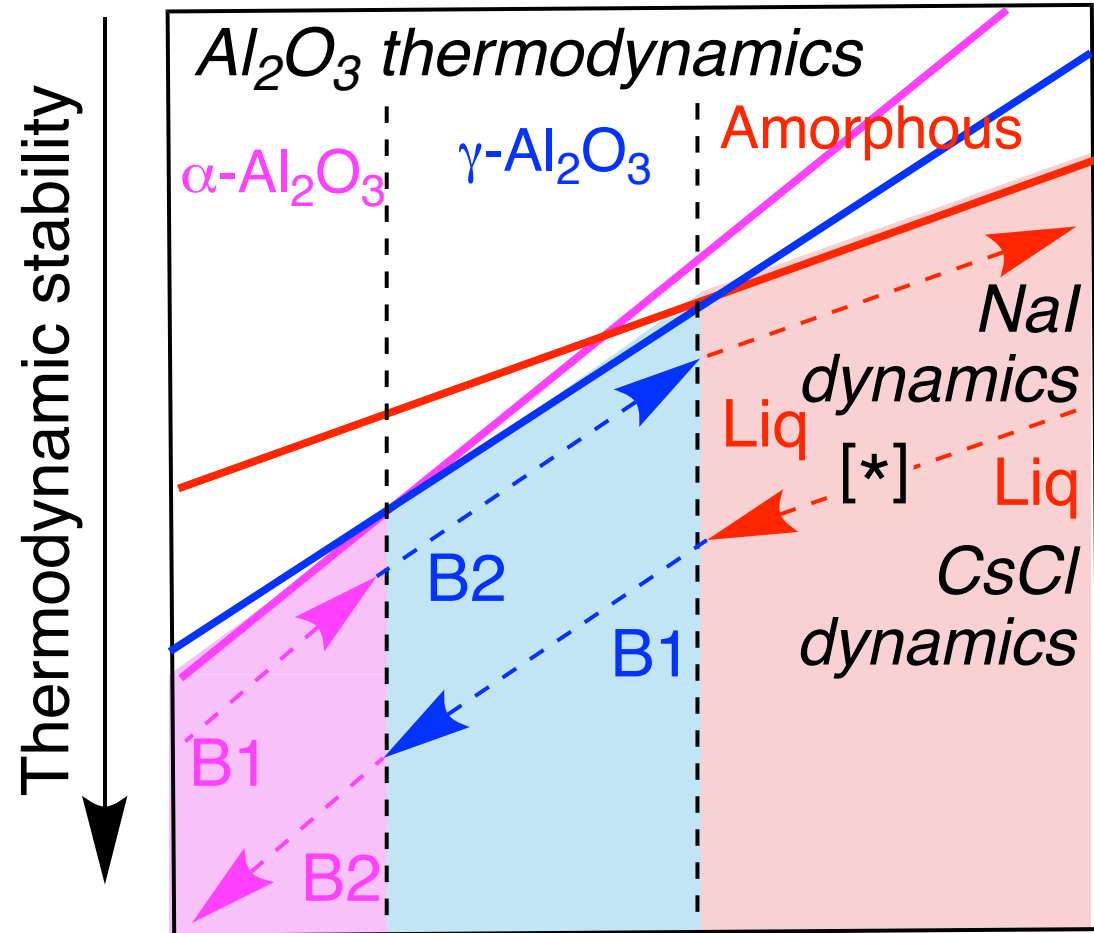
Presented in the pre-symposium

T. Nakamuro *et al.* *J. Am. Chem. Soc.* **2021**, 143, 1763.

M. Sakakibara *et al.* *ACS Cent. Sci.* **2022**, 8, 1704.

Topic: Size-dependent Polymorphism

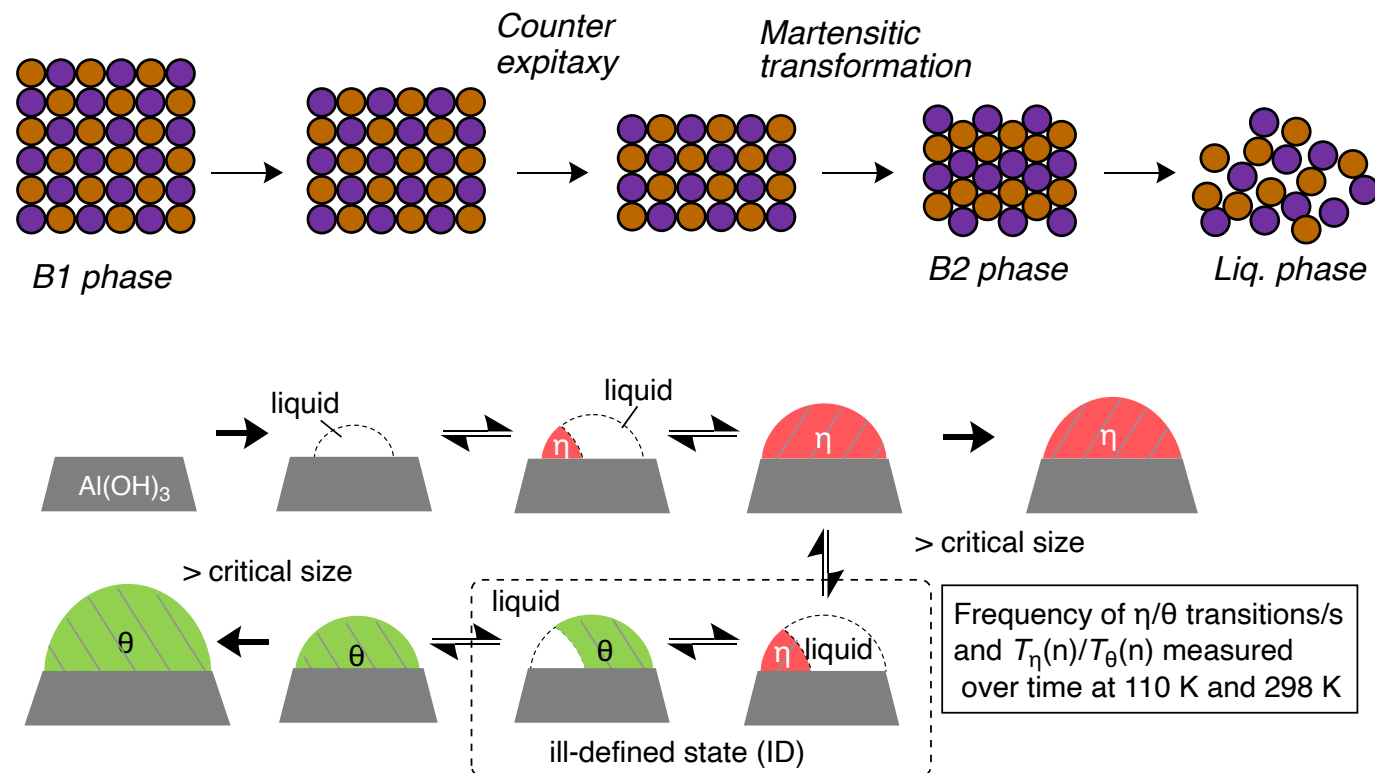
diameter = ~ 13.5 nm ~ 4.5 nm



diameter = ~ 3 nm ~ 2.5 nm

Surface area

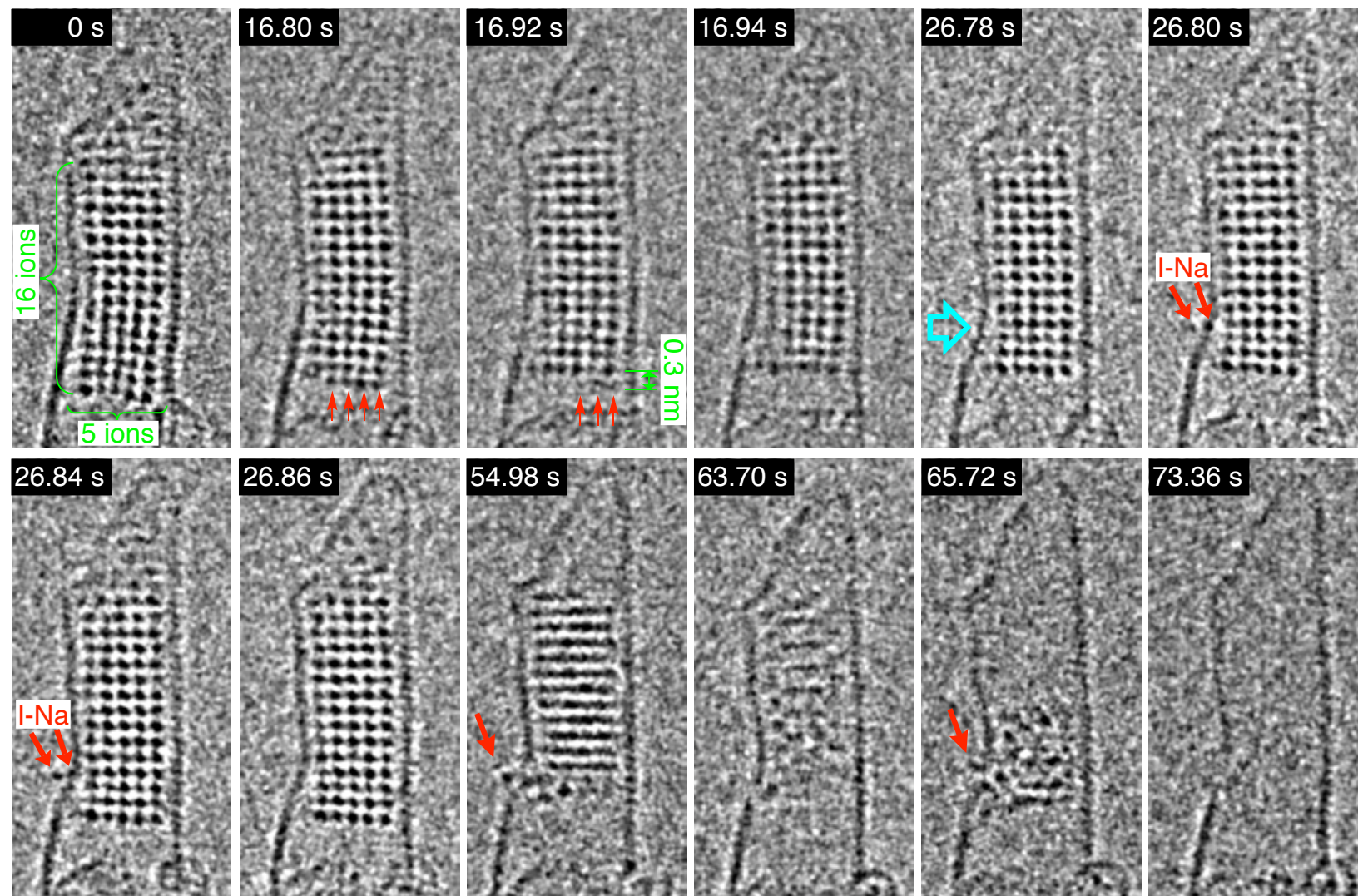
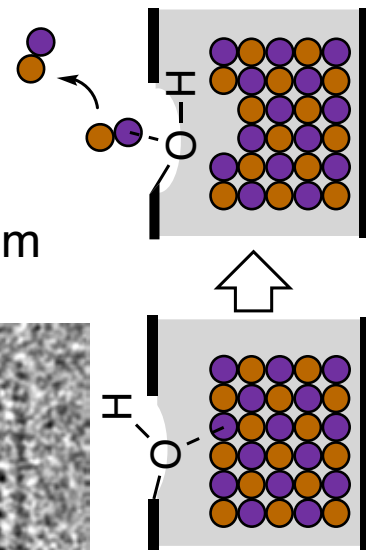
Particle size



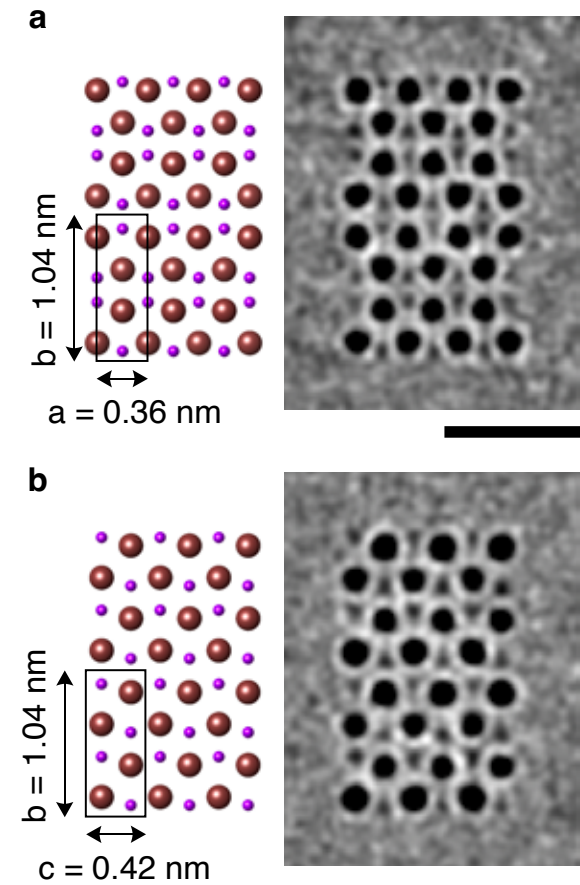
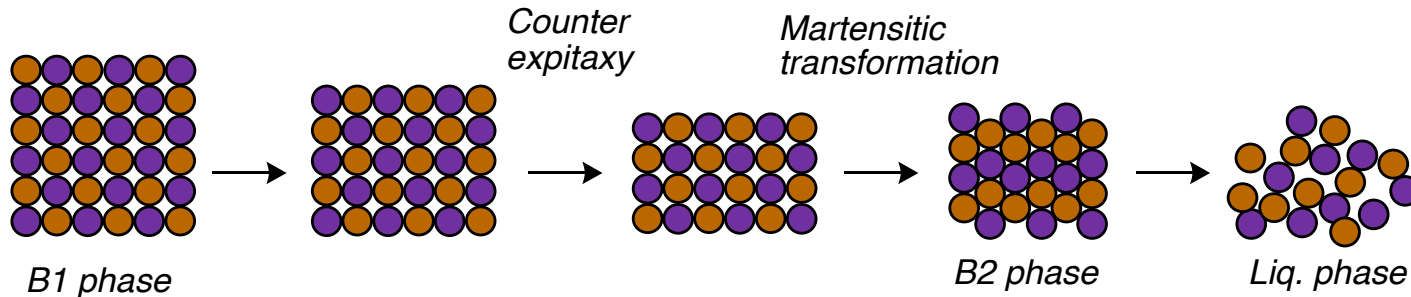
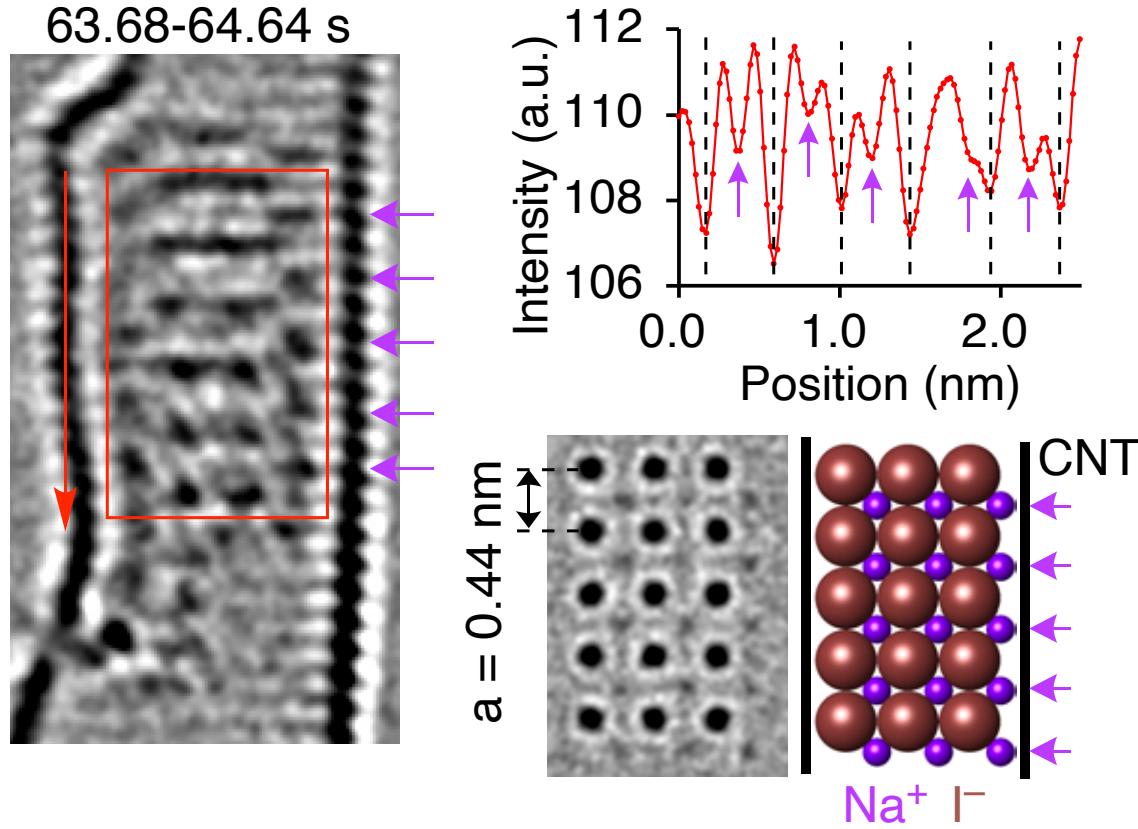
Additional: Micro-meter world

Search for Polymorphs Using *Nanotest Tube Method*

50 fps, EDR of $4.0 \times 10^5 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$, 298 K, vacuum



Structural Analysis of Unknown Polymorph

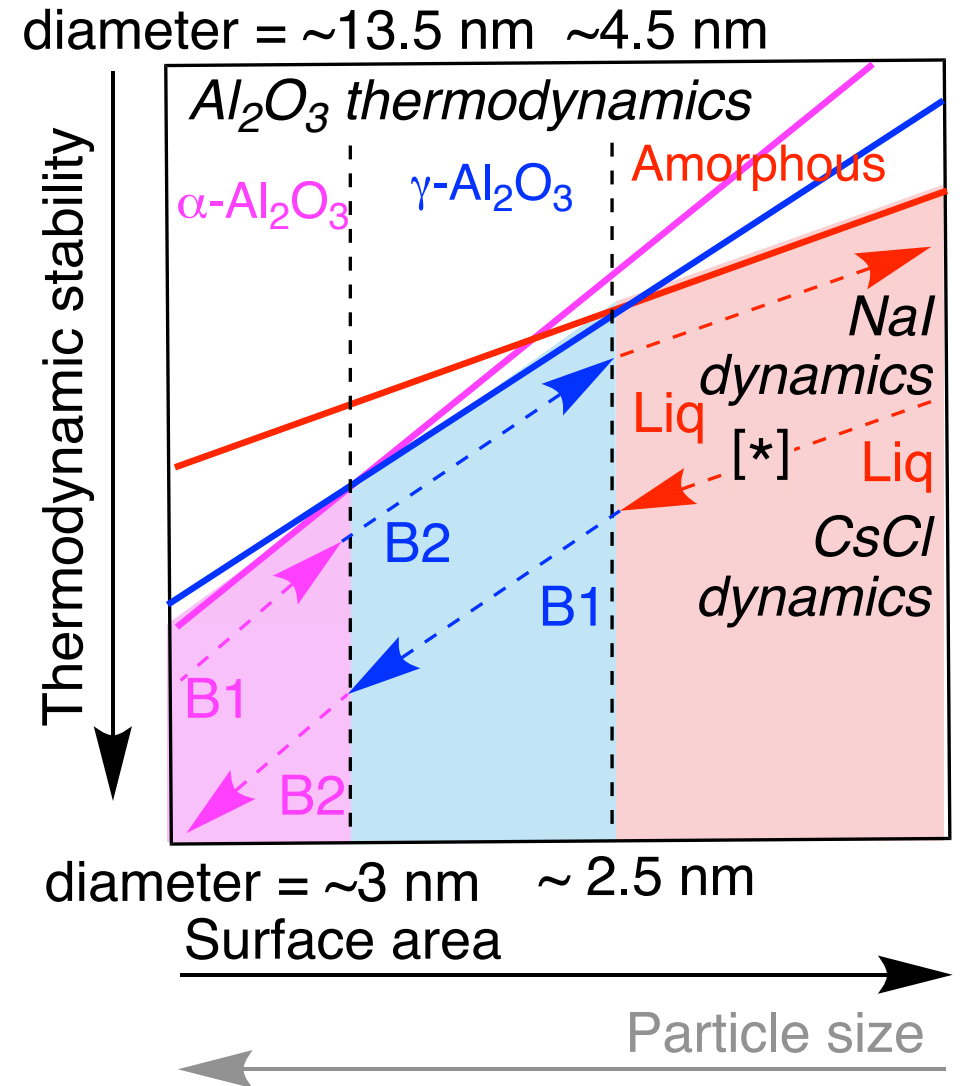
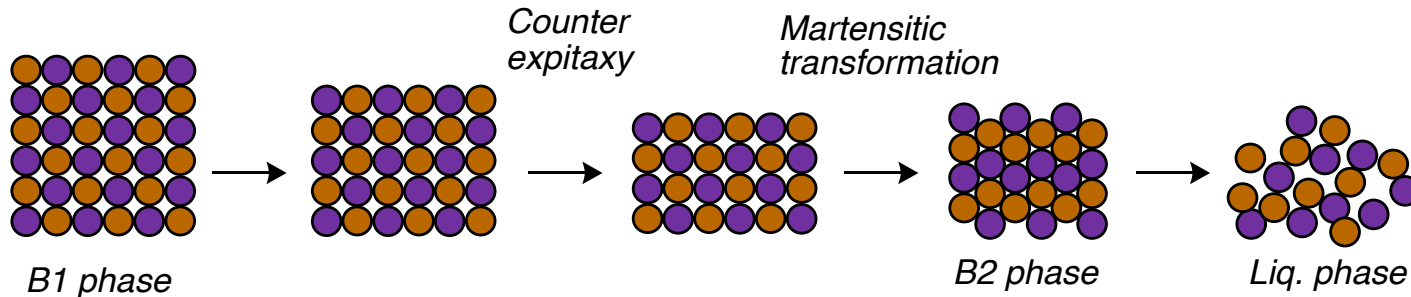
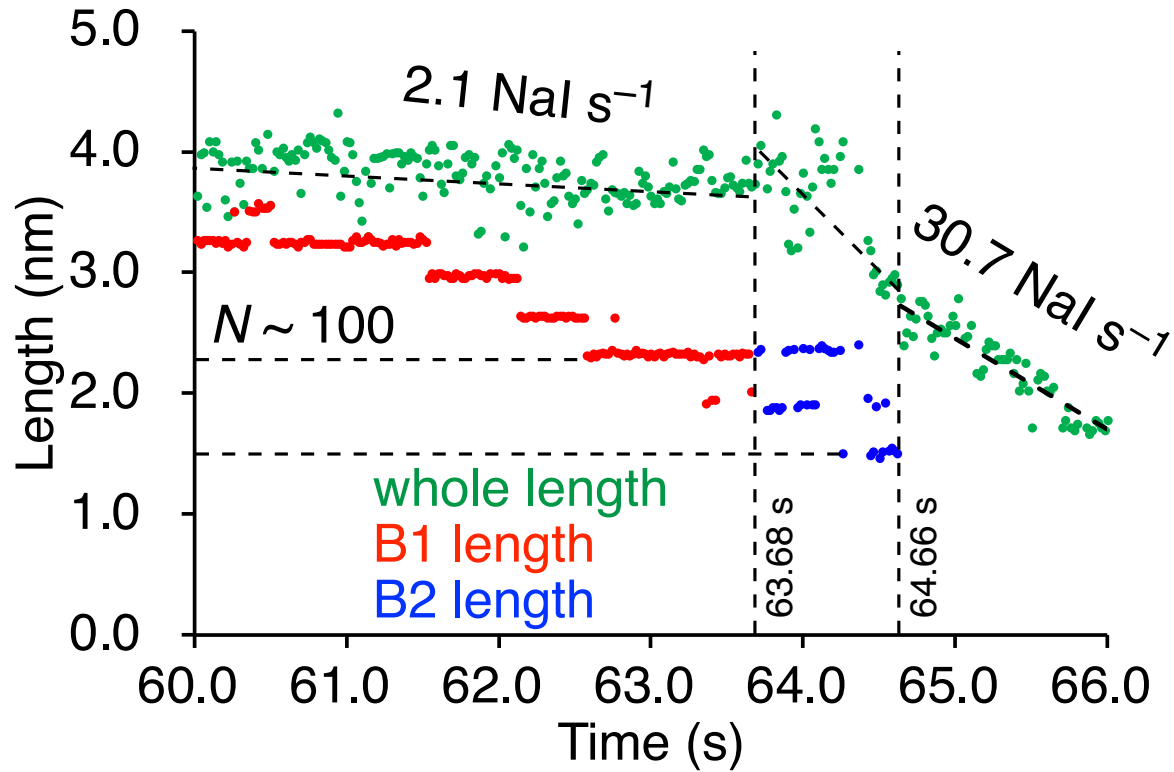


J. M. Léger *et al.*, *J. Phys.: Condens: Matter* **1998**, 10, 4201.
(B33 structure at 40 GPa)

A. Navrotsky *et al.*, *Science* **1997**, 277, 788.

M. Sakakibara *et al.*, *ChemRxiv* (10.26434/chemrxiv-2024-ms4t7)

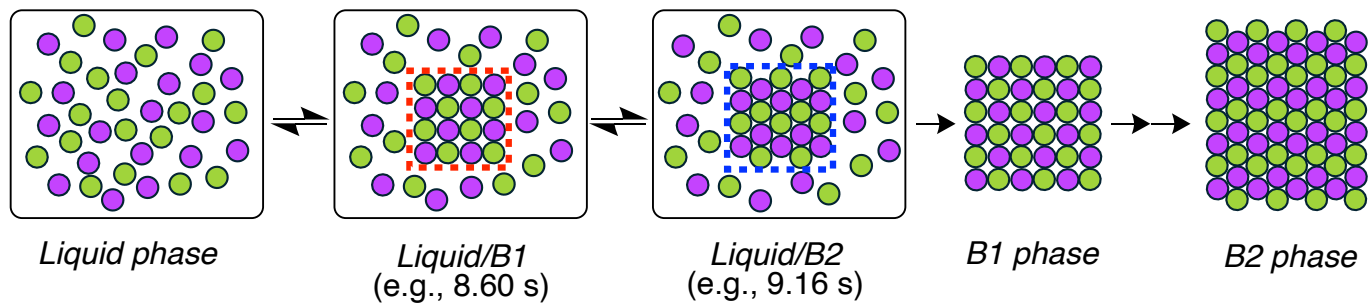
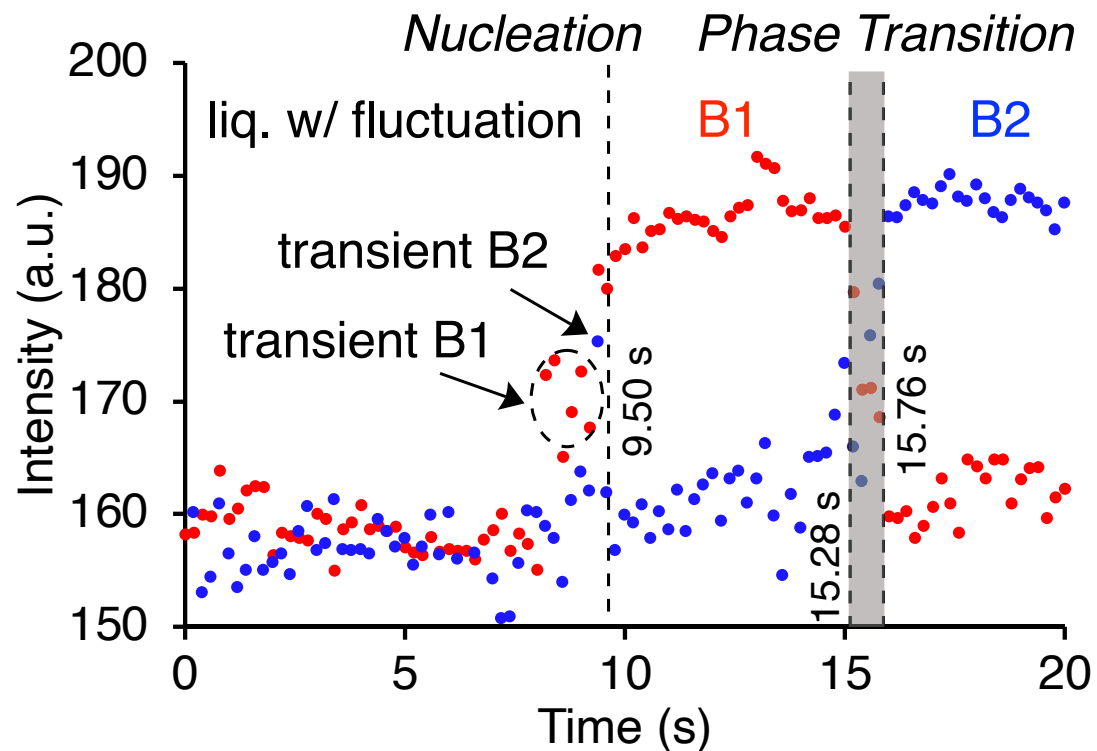
Stability of Nanocrystalline Polymorphs Reversed



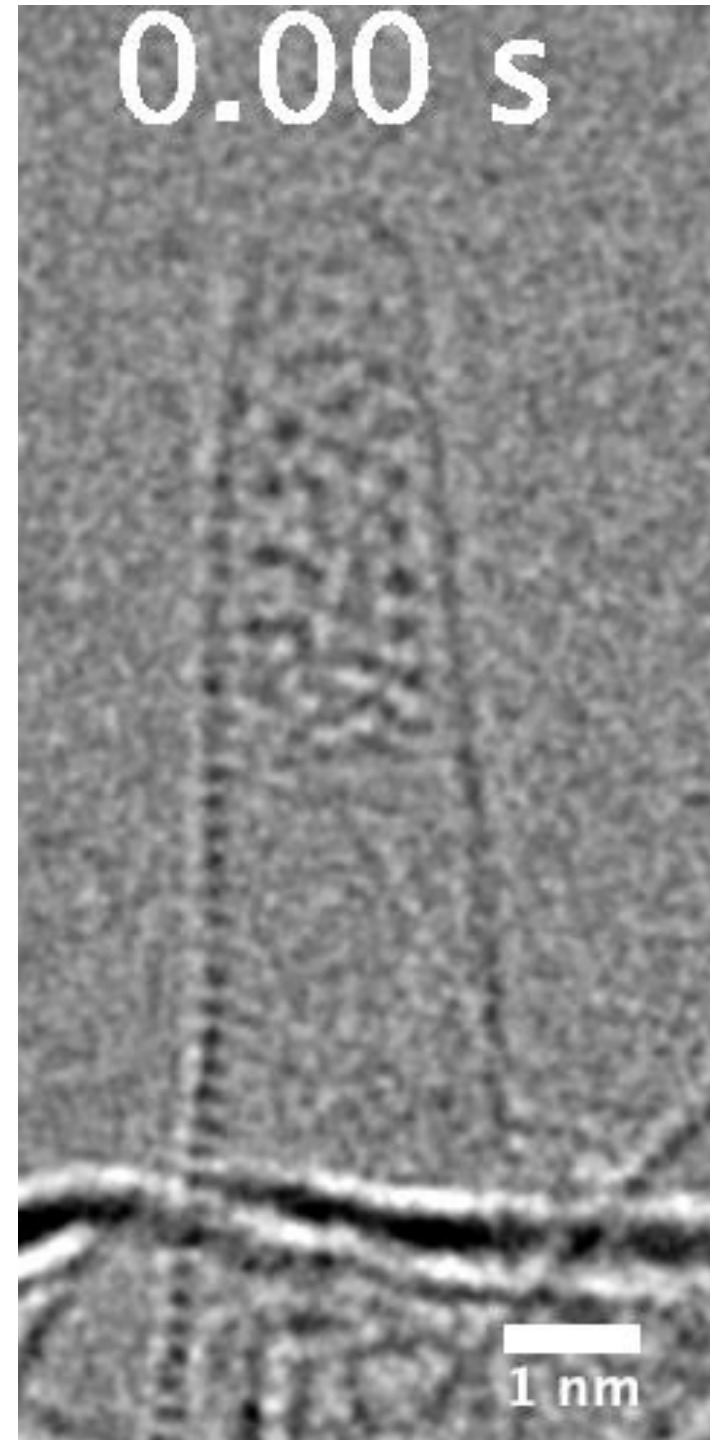
A. Navrotsky *et al.*, *Science* **1997**, 277, 788.

M. Sakakibara *et al.*, *ChemRxiv* (10.26434/chemrxiv-2024-ms4t7)

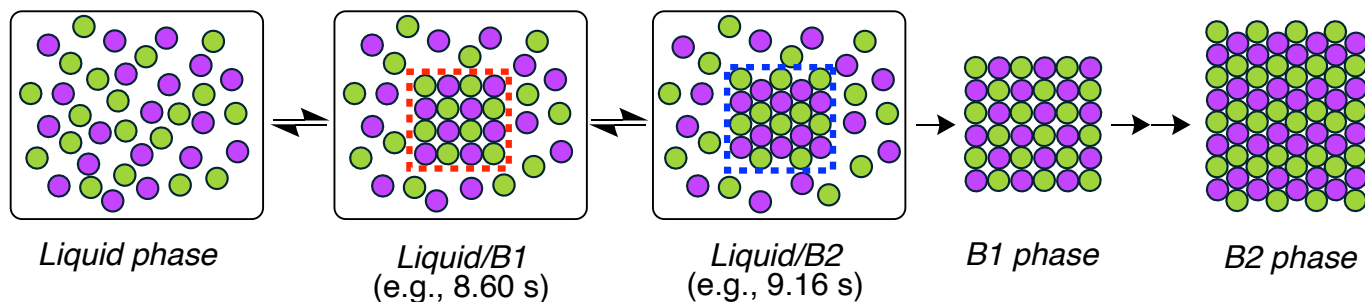
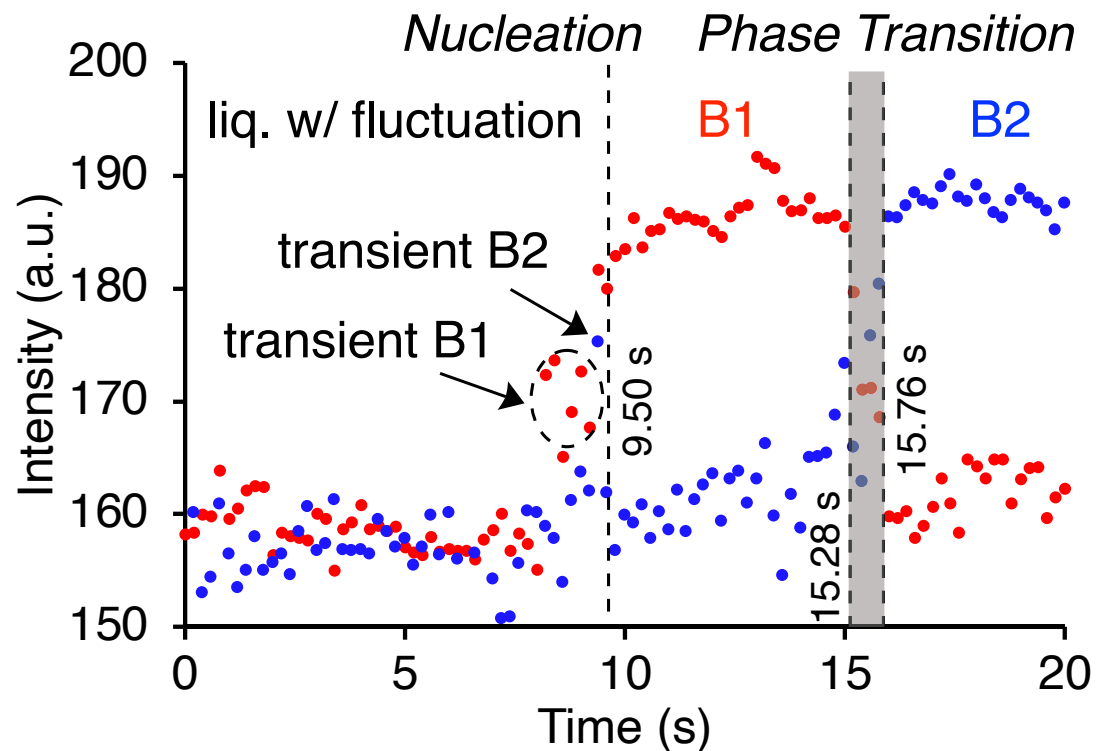
CsCl Nucleates as NaCl-type



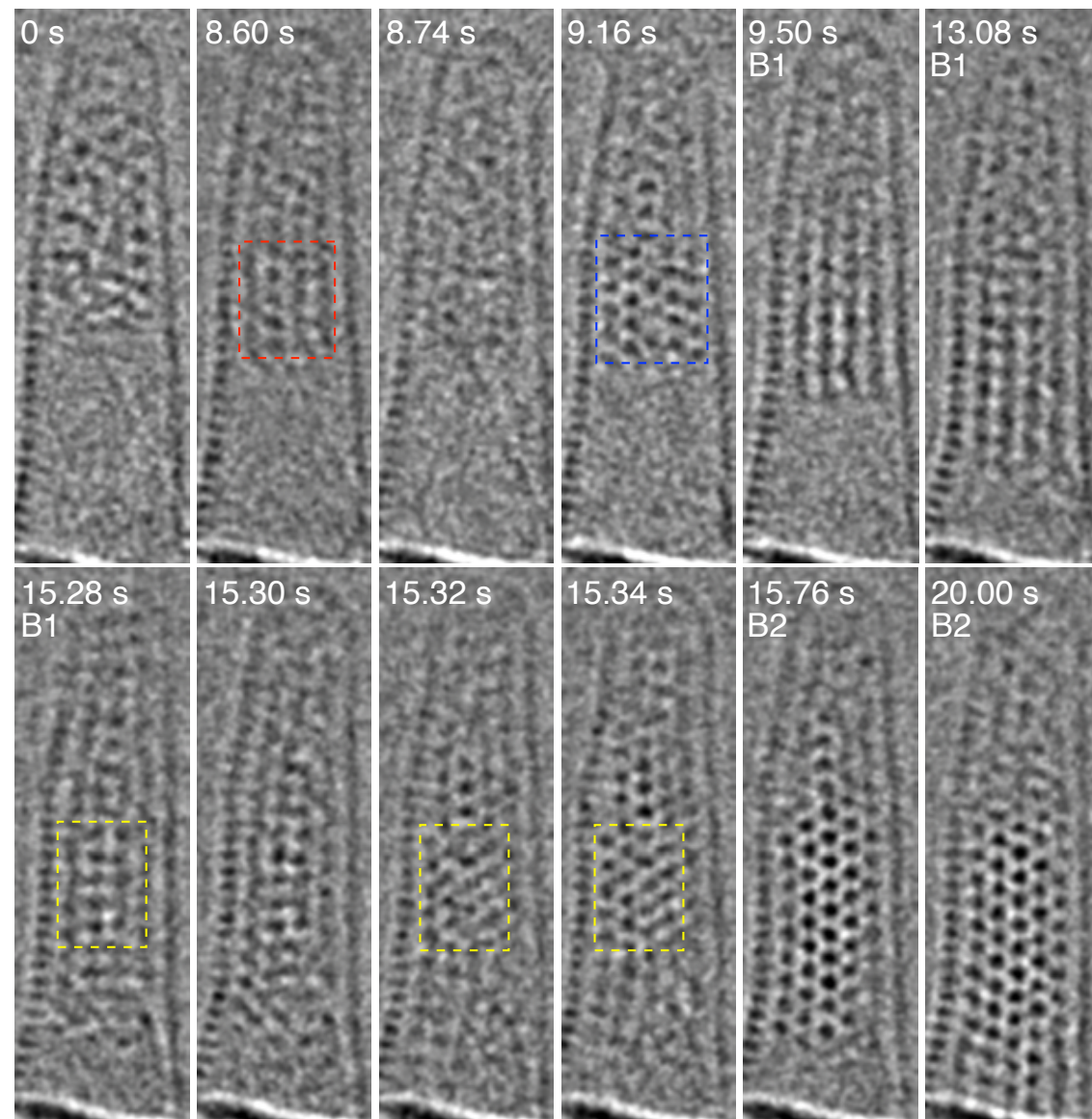
50 fps, EDR of $3.4 \times 10^6 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$
 298 K, vacuum



CsCl Nucleates as NaCl-type

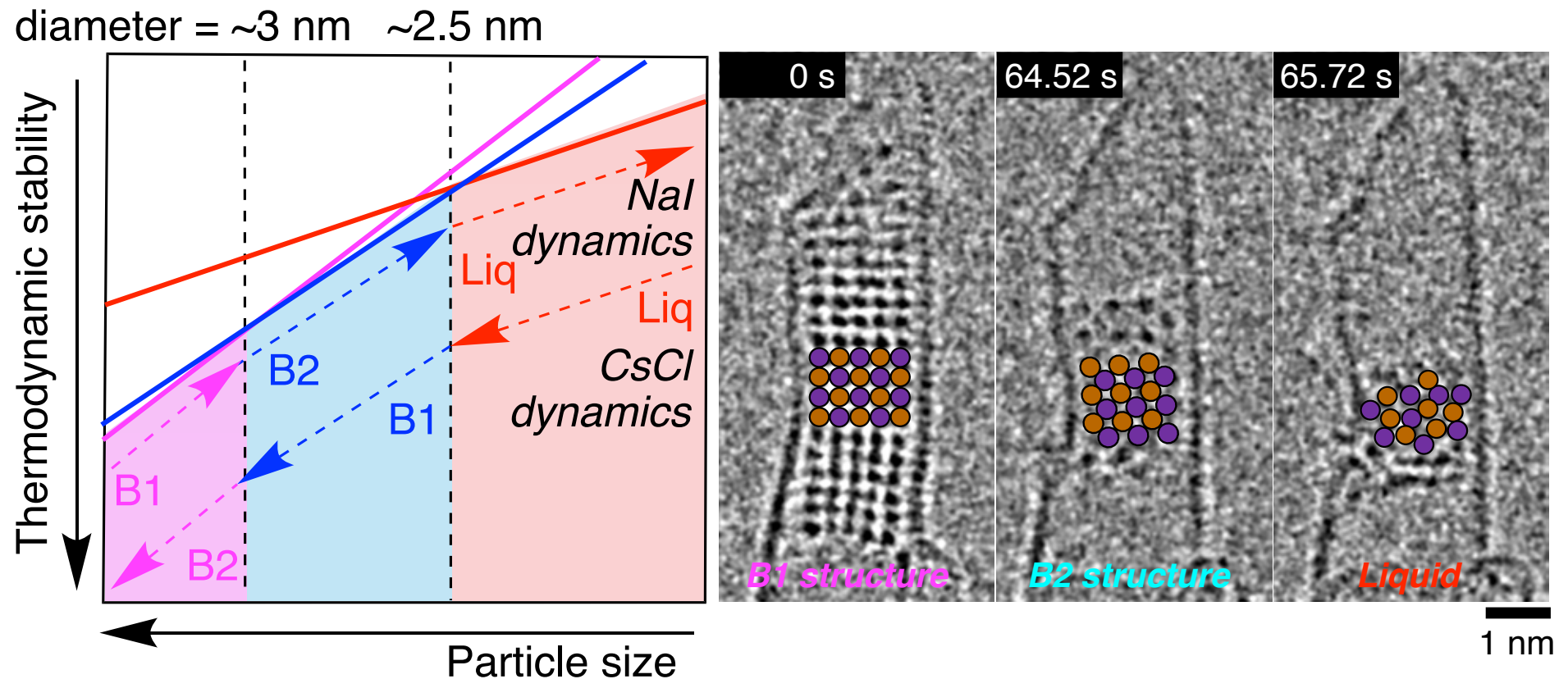


50 fps, EDR of $3.4 \times 10^6 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$
 298 K, vacuum

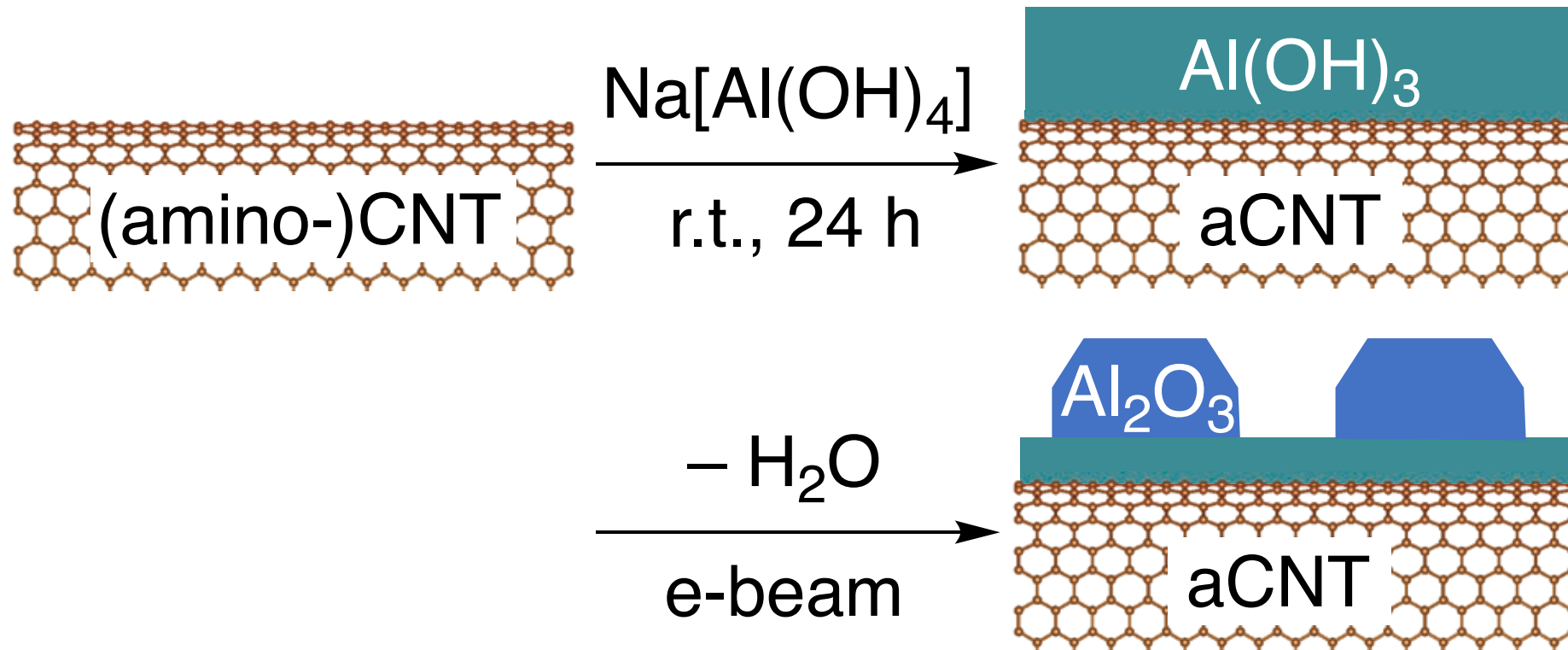


Summary of This Topic

- TEM direct imaging of nucleation/growth of nanocrystals enabled the exploration of its mechanism and size dependency in polymorphism.
- Nano-specific polymorphs can give us important insights for designing nanomaterials with unique structures.



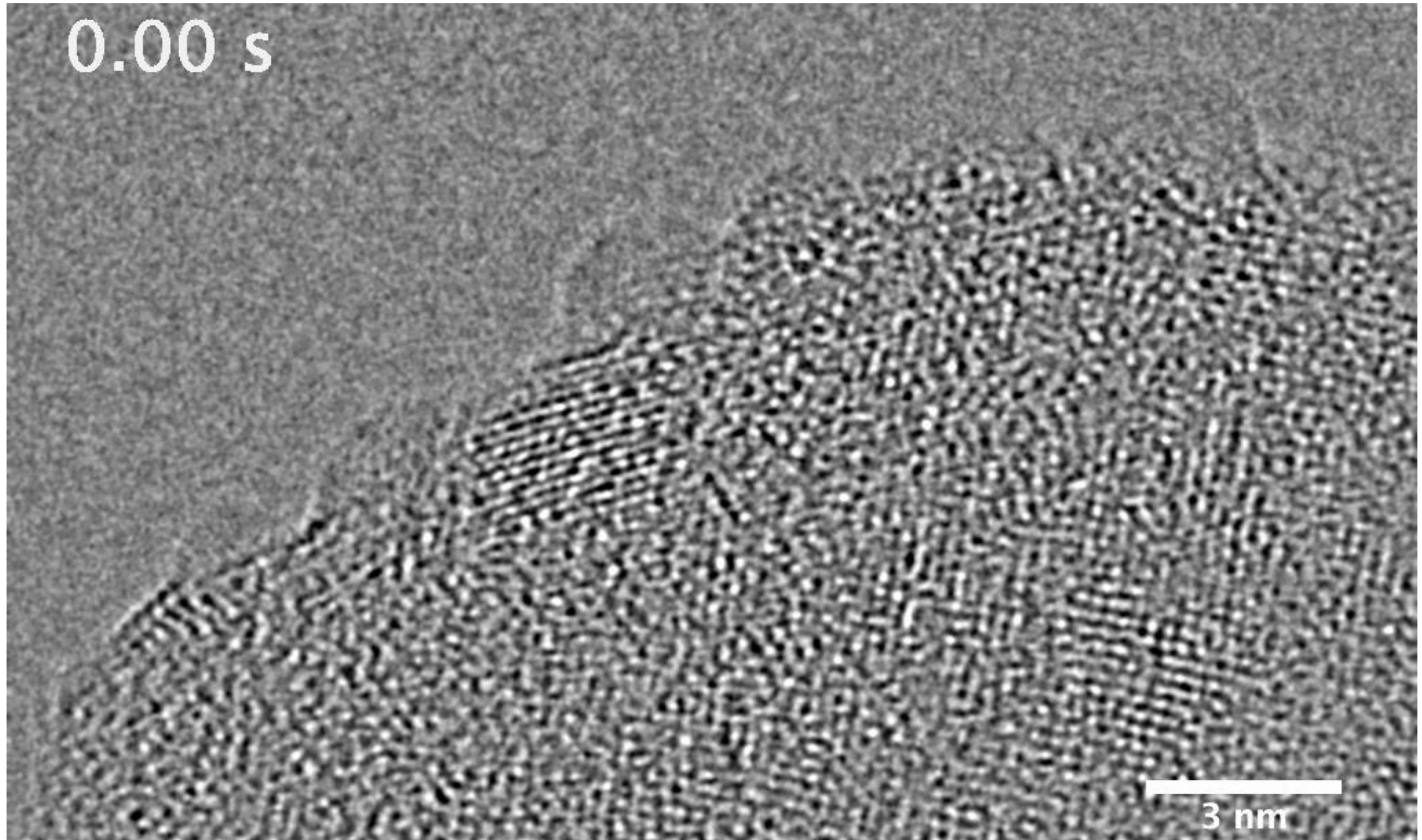
Chemical Coating: Unrestricted Observation Space



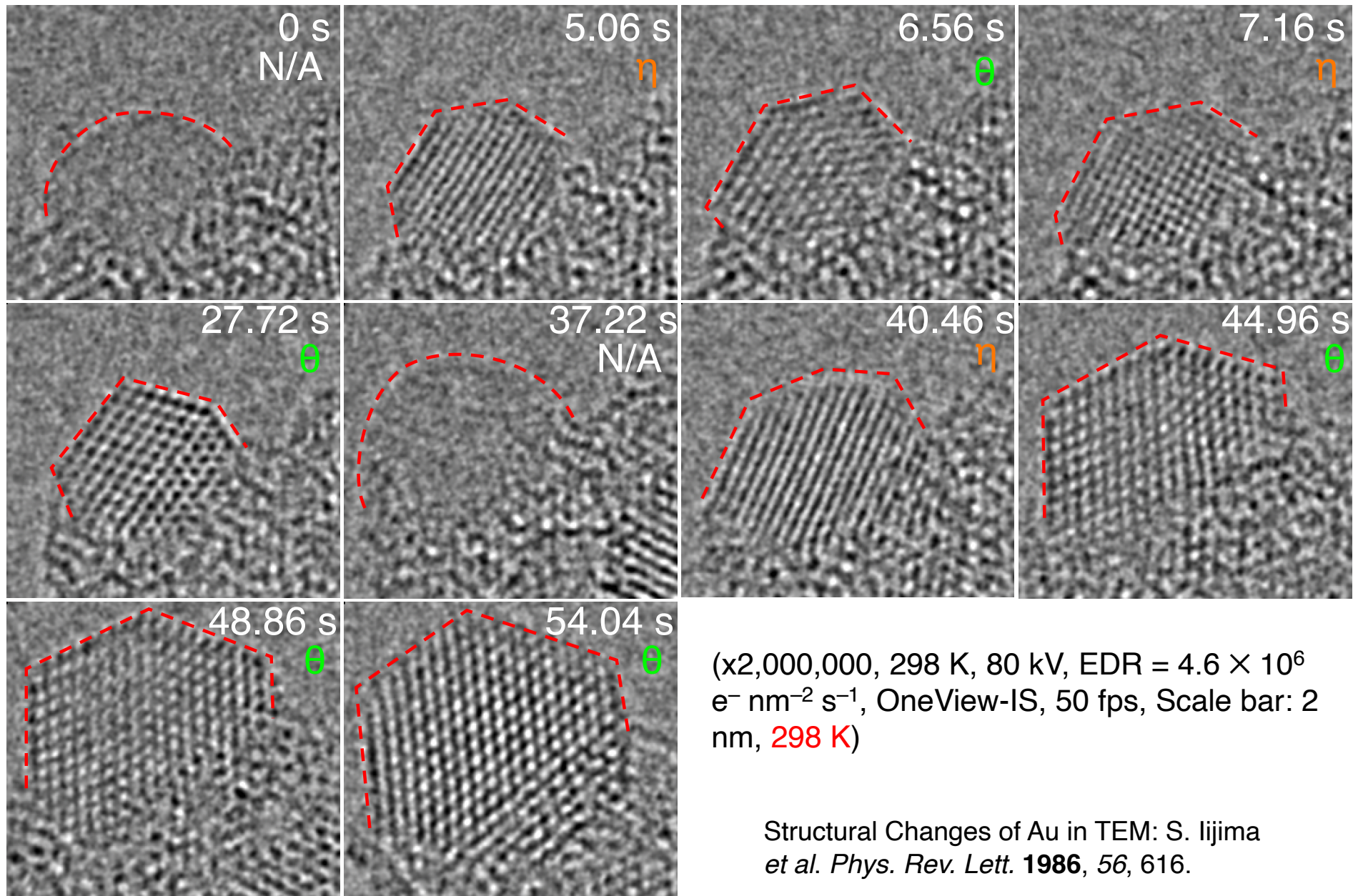
Analyzed by PXRD and TG-DTA

Insoluble $\text{Al}(\text{OH})_3$ was deposited onto CNTs through hydrolysis & heterogeneous nucleation.

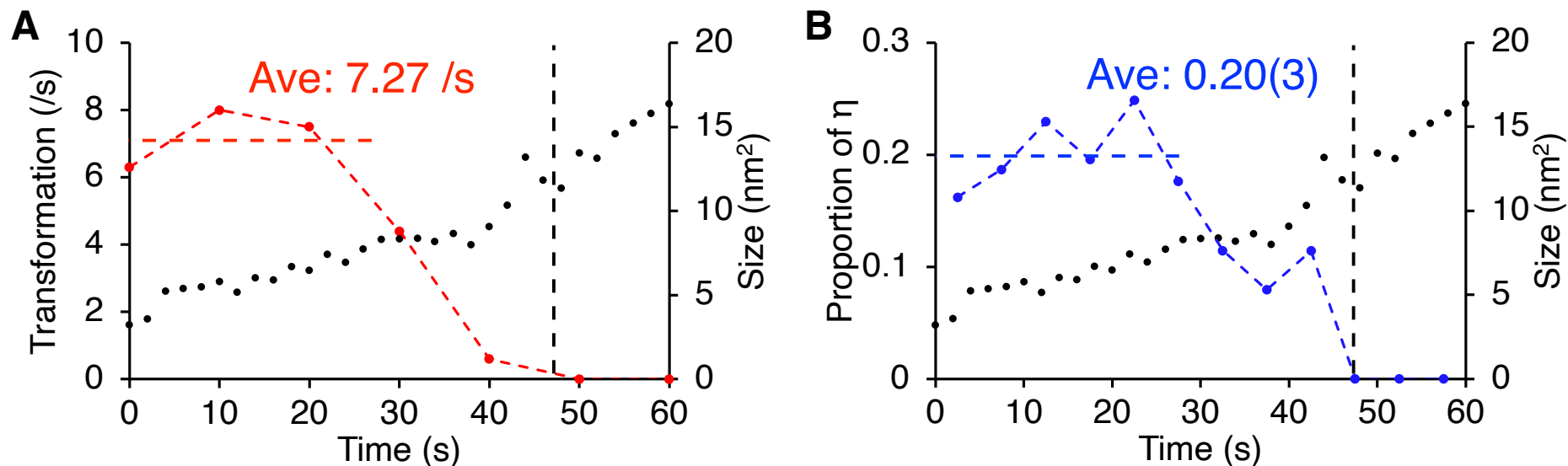
Al_2O_3 Formation under TEM Observation



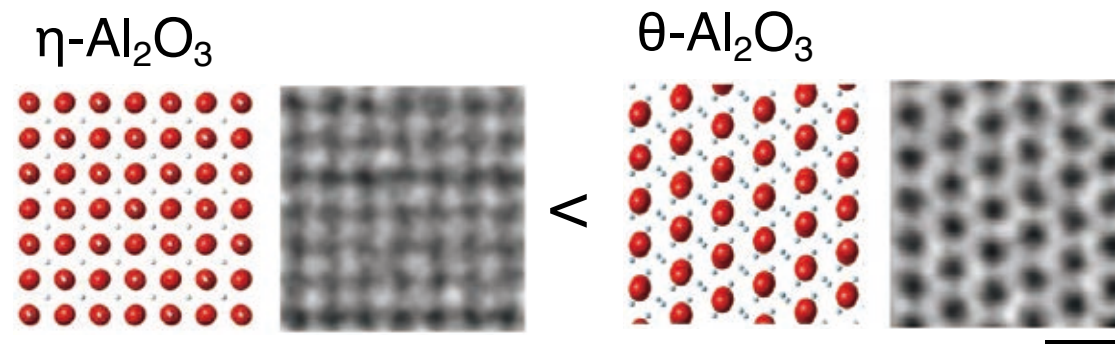
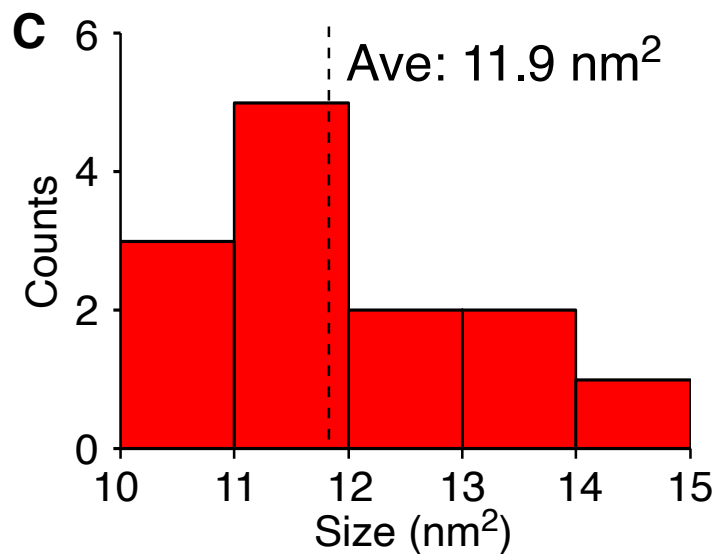
Polymorphism Is Not *Deterministic* But *Stochastic*



Time-Evolution of Structural Features



(298 K)



Stabler θ -Al₂O₃ became dominant in the late stage of nucleation.

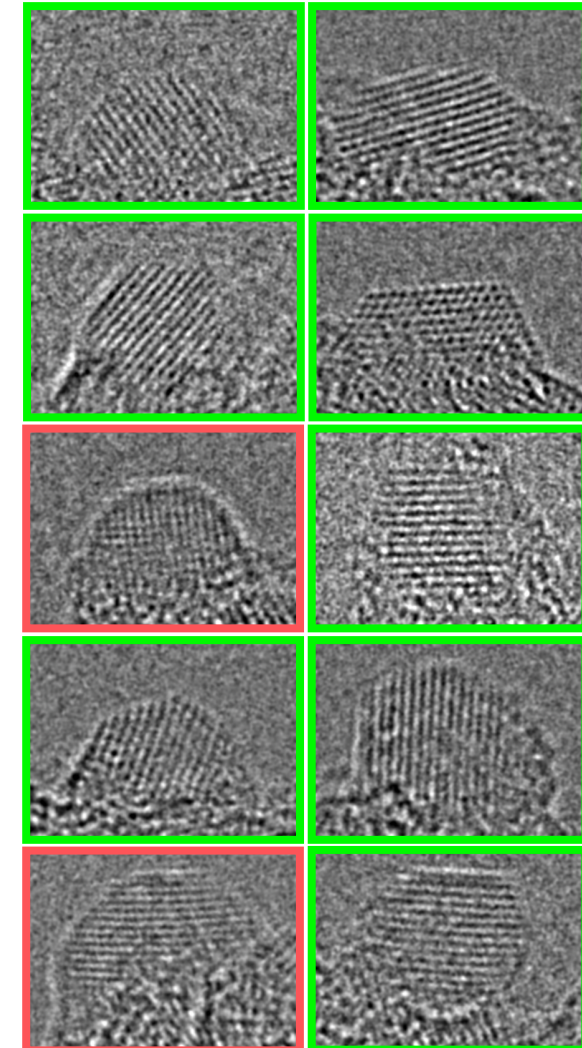
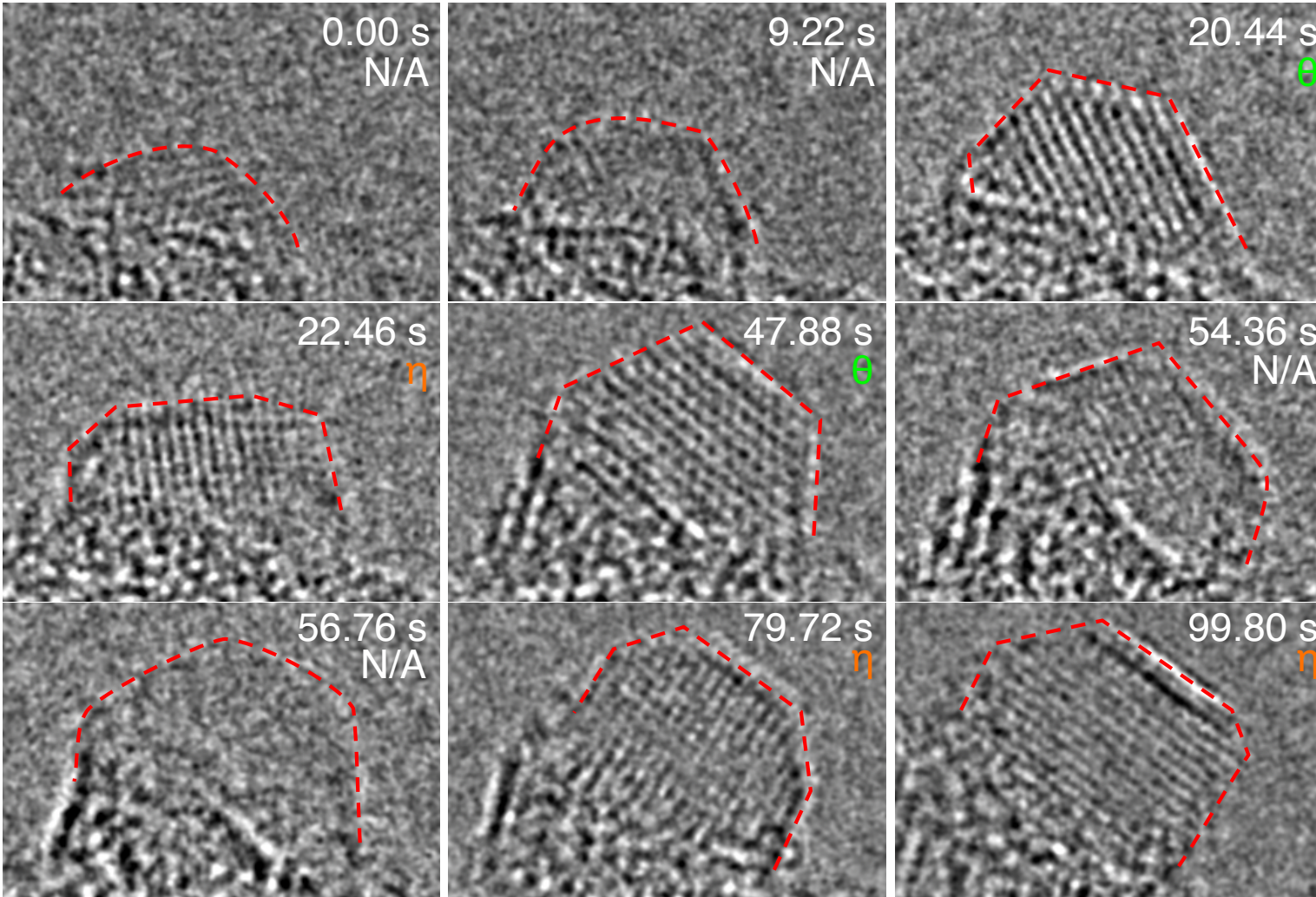
Fluxionality Frozen at Low Temperature

$$r_{\text{c,melt,hets}} = \frac{2\gamma T^*}{\Delta H_f \Delta T}$$

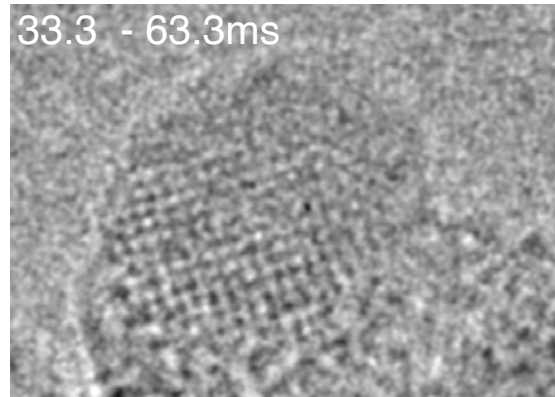
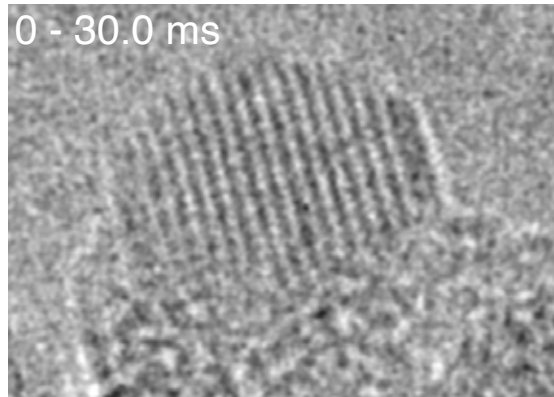
T^* : melting point, T : Experimental Temp.

$$\Delta T = T^* - T$$

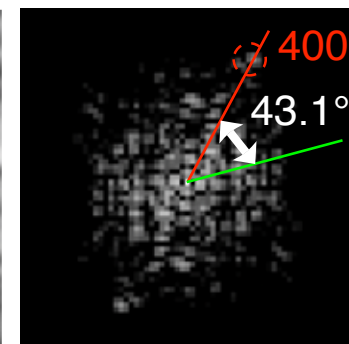
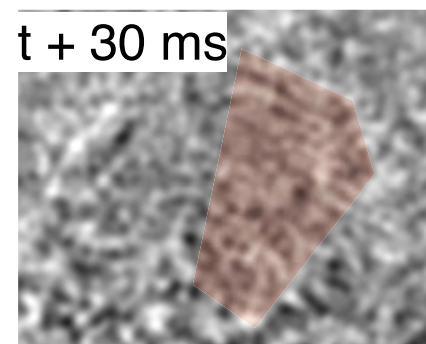
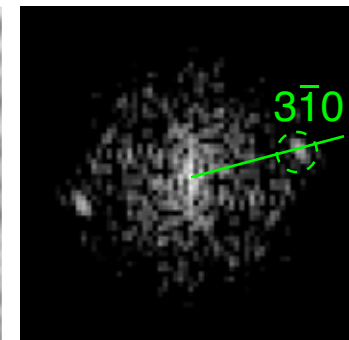
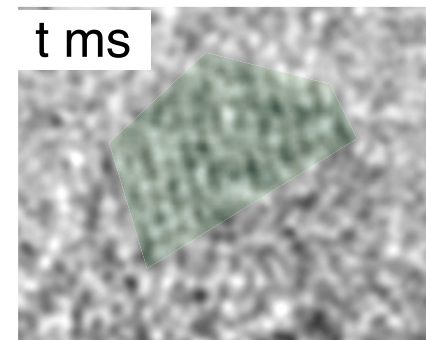
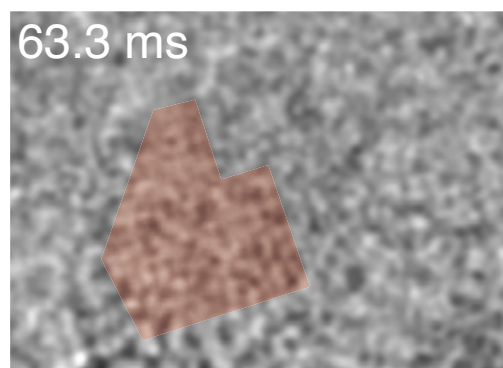
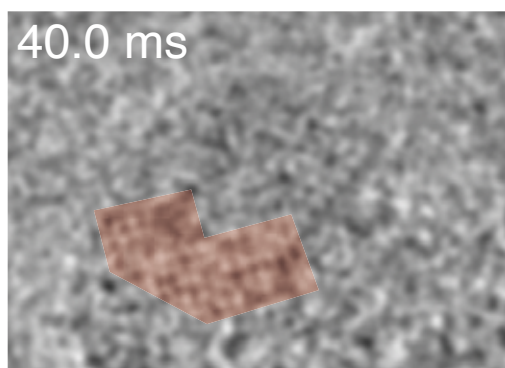
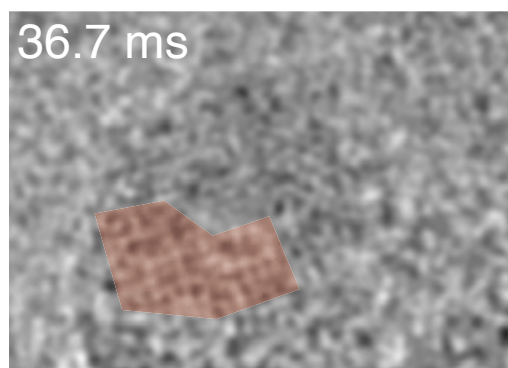
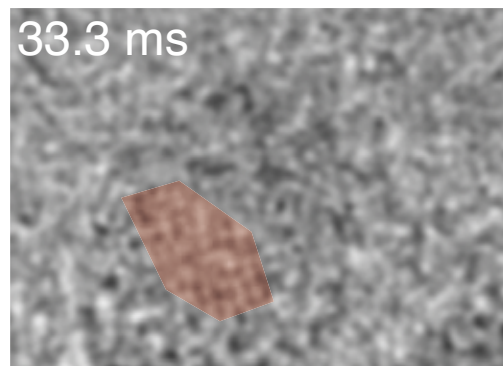
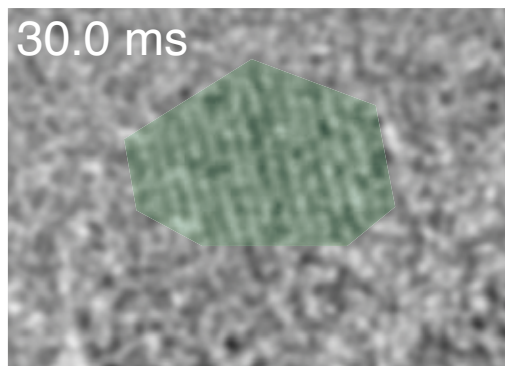
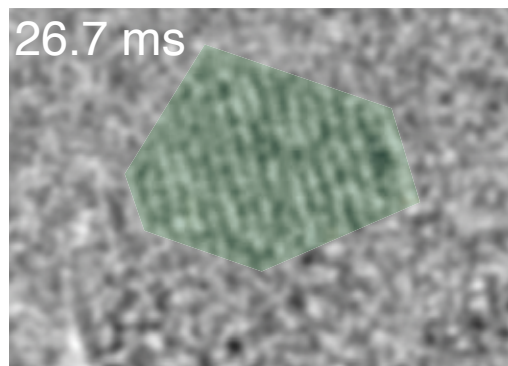
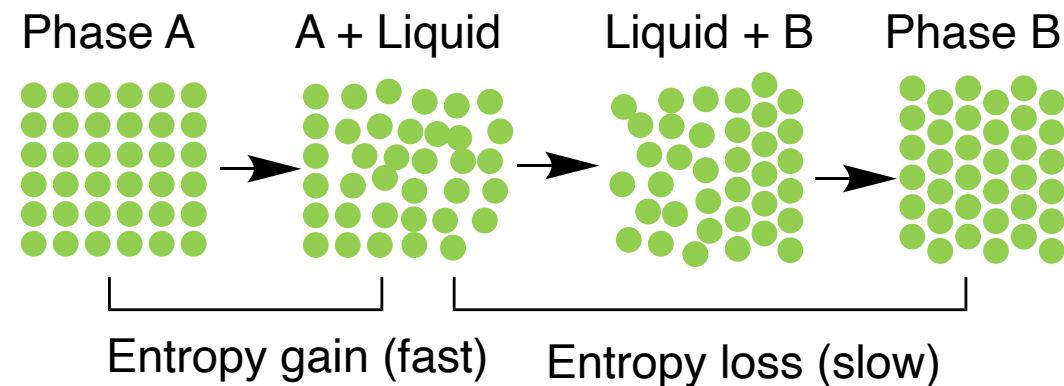
(at 110 K)



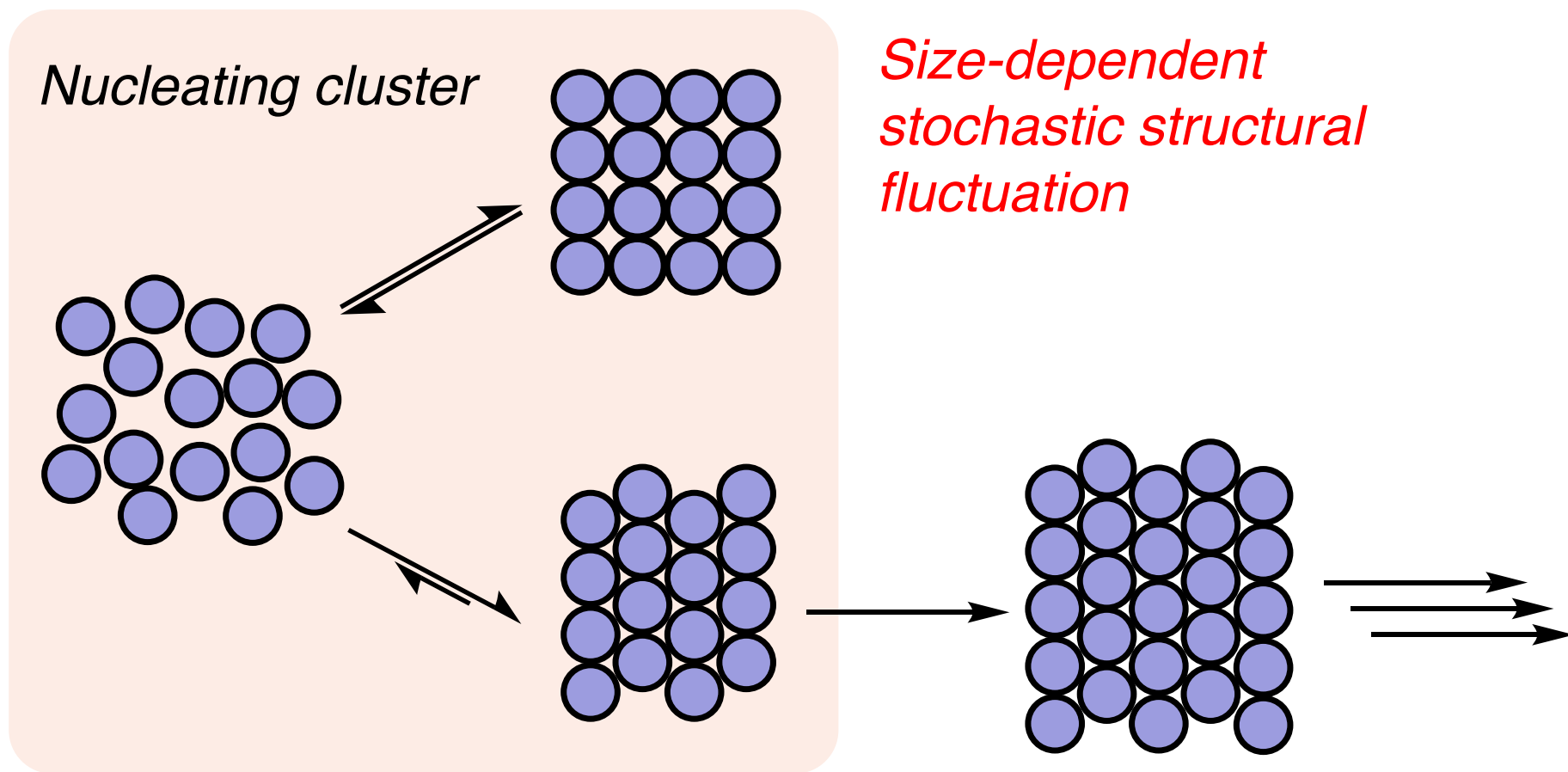
Polymorphic Transition by Fast Observations



(K3-IS, 300 fps, at 298 K)



Fluxionality during Crystallization



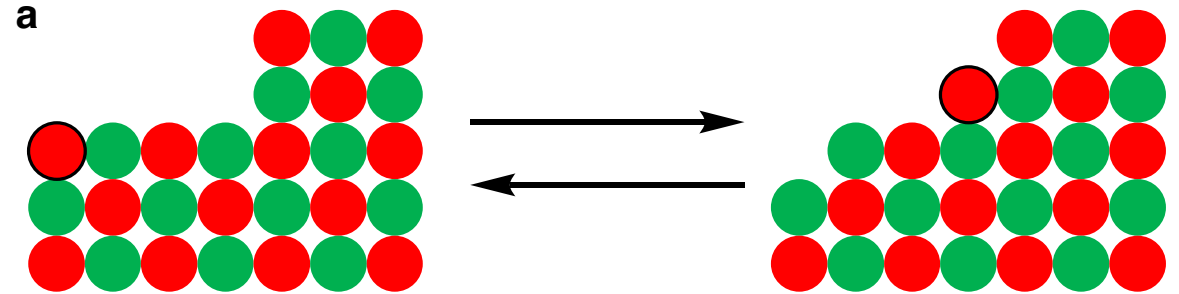
M. Sakakibara *et al.*, *ChemRxiv* (10.26434/chemrxiv-2024-g8tl8)

Fluxionality

V. Fung, D.-e. Jiang, *J. Phys. Chem. C* **2017**, 121, 10796.

Z. Zhang, B. Zandkarimi, A. N. Alexandrova, *Acc. Chem. Res.* **2020**, 53, 447.

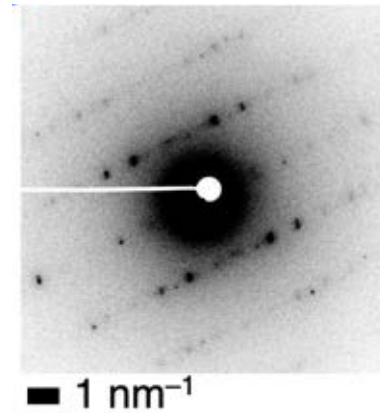
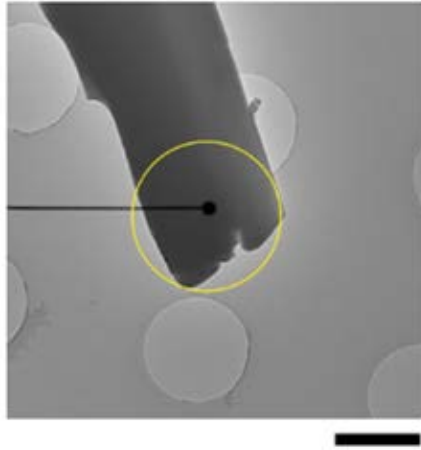
Sub-micrometer: Uniformization of Terrace-Width on CaO



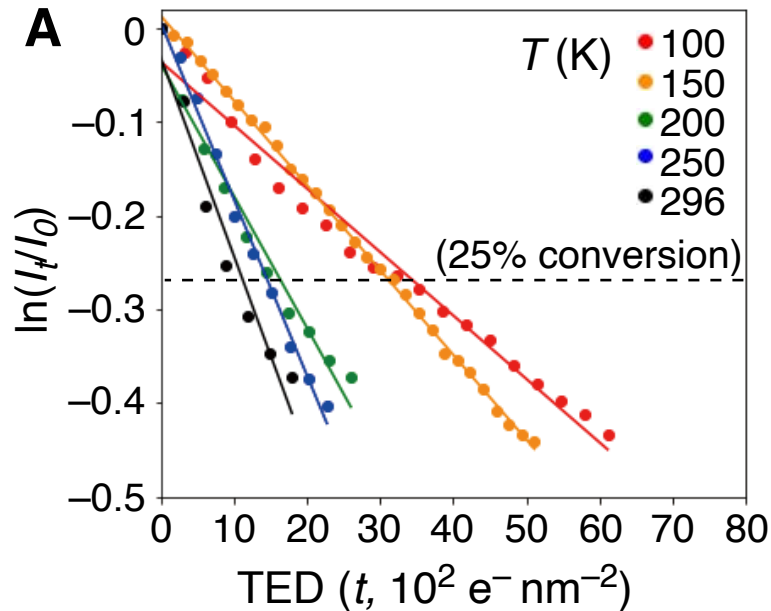
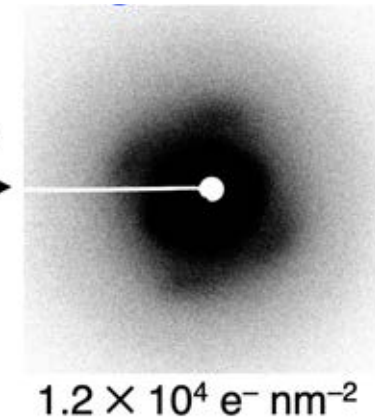
EDR = $6.5 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$. OneView-IS.
50 fps. Scale bar: 3 nm. After BP filtering.
RT.

unpublished

Micrometer: Excitation Causes Molecular Disordering in Lattices by ED

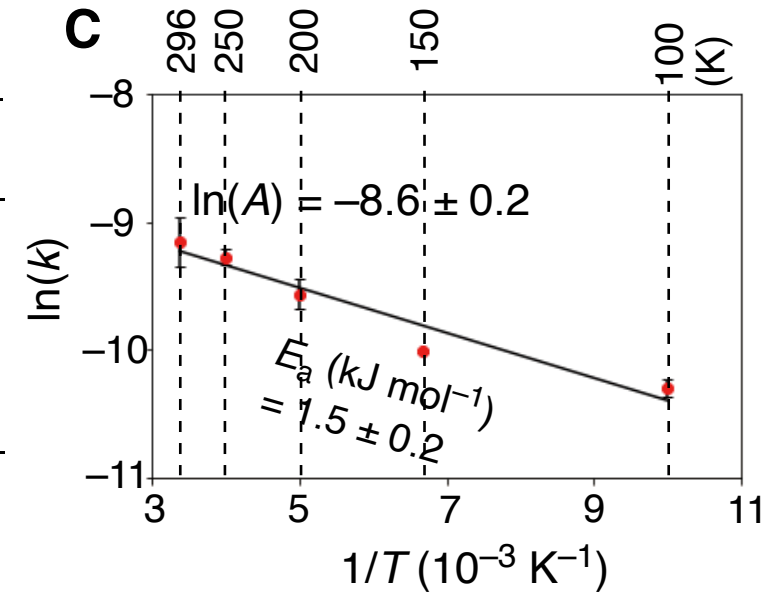


e-beam



B

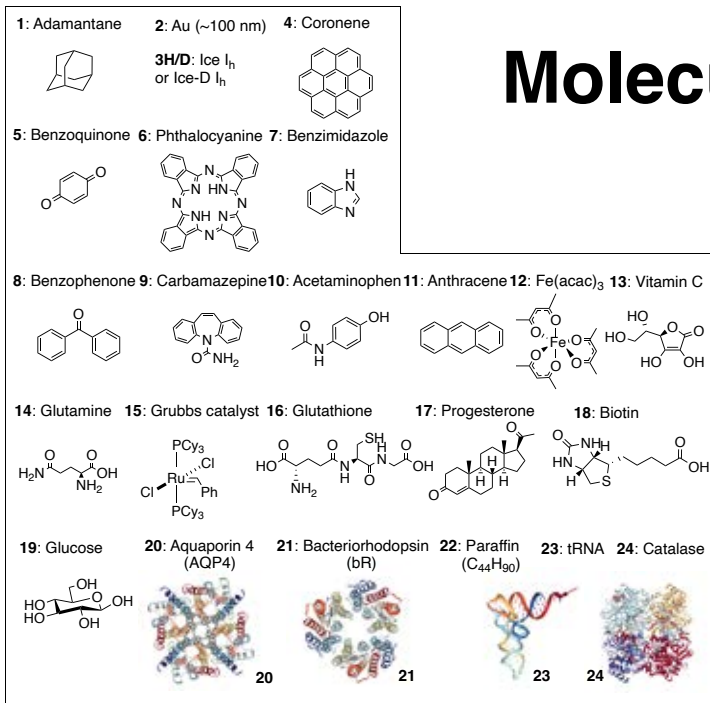
$T \text{ (K)}$	$k \text{ (} 10^{-4} \text{ (e}^-)^{-1} \text{ nm}^2 \text{)}$
100	0.34 ± 0.06
150	0.45 ± 0.01
200	0.70 ± 0.08
250	0.94 ± 0.05
296	1.1 ± 0.2



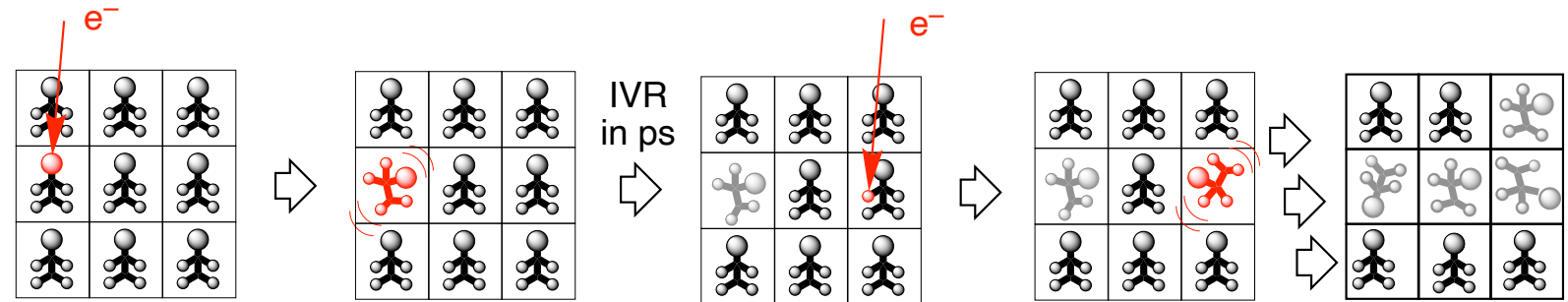
Diffraction intensity $I \propto f^2 \underline{N_{UC}}^2$
 (f : structure factor, $\underline{N_{UC}}$: number of UCs)



$\ln(N_t/N_0) = -k \text{ TED}$,
 TED: total electron dose, N_t : number of ordered molecules

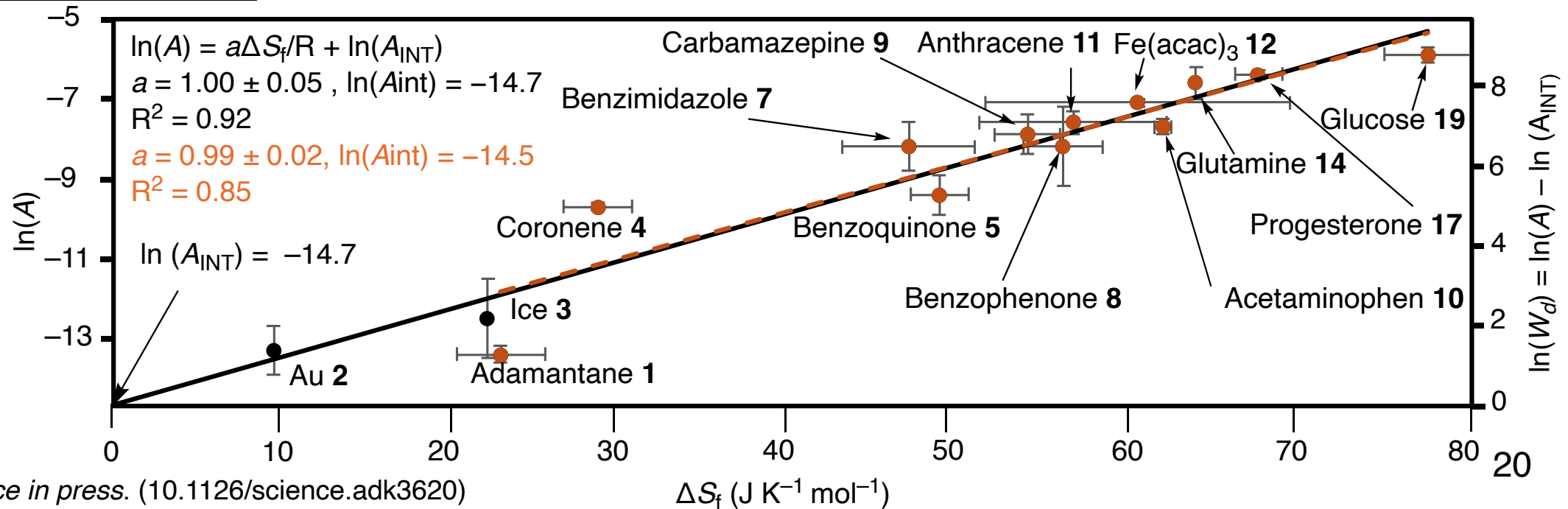


Molecular Disordering Corresponds to Melting in Bulk



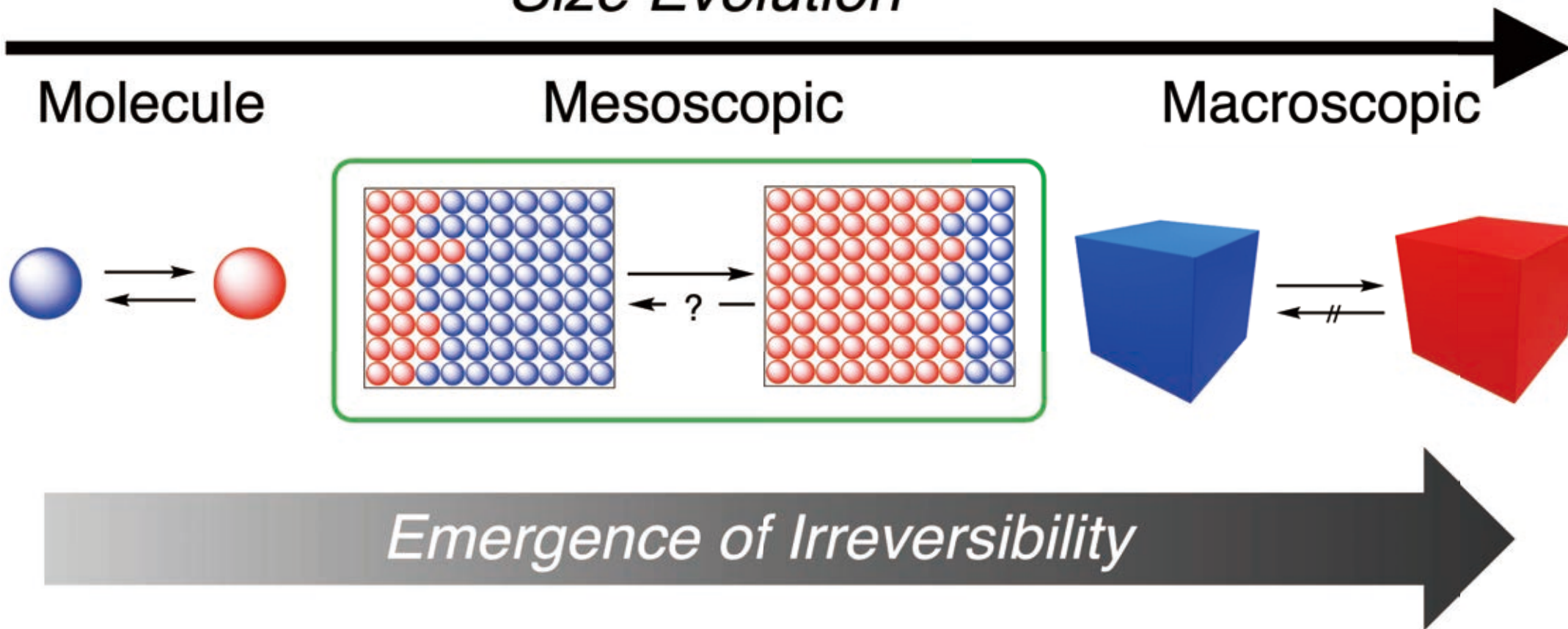
200 keV : $\lambda = 2.5$ pm, EDR : $\sim 10^3$ e⁻ nm⁻² s⁻¹, T : 100-296 K

(IVR: Intramolecular Vibrational energy Redistribution)



Conclusion & Perspectives

Size Evolution



Establishment of "*Cinematic Chemistry*" will allow us to experimentally explore the relation between deterministic chemical processes in molecular assembly and the stochastic behavior of individual atoms/molecules.

Acknowledgements



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Dr. Dongxin Liu
Mr. Jiarui Fu
Prof. Minoru Hanaya@Gunma U.

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- JST PRESTO



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